8(14),15-SANDARACOPIMARADIENE-2α,18-DIOL, A MINOR CONSTITUENT OF THE RWANDESE MEDICINAL PLANT TETRADENIA RIPARIA

Luc Van Puyvelde,* Norbert De Kimpe,*† François Borremans,‡ Weiguo Zhang‡ and Niceas Schamp†

CURPHAMETRA (Centre Universitaire de Recherches sur la Pharmacopée et la Médecine Traditionelle), B.P. 117, Butare, Rwanda, Central Africa; † Laboratory of Organic Chemistry, Faculty of Agricultural Sciences, State University of Gent, Coupure Links 653, B-9000 Gent, Belgium; ‡ Laboratory of Organic Chemistry, State University of Gent, Krijgslaan 281, B-9000 Gent, Belgium

(Received 25 April 1986)

Key Word Index—Tetradenia riparia; Labiatae; diterpenediol; 8(14),15-sandaracopimaradiene-2α,18-diol.

Abstract—An investigation directed towards minor constituents of *Tetradenia riparia* (previously named *Riboza riparia*) resulted in the isolation of 8(14),15-sandaracopimaradiene- $2\alpha,18$ -diol, a new natural diterpenediol.

INTRODUCTION

Recently [1], the active principle of *Tetradenia riparia* (Hochst) Codd. (previously named *Iboza riparia*), an important plant in the traditional medicine of Rwanda (Central Africa), was identified as 8(14),15-sandaracopimaradiene-7α,18-diol (1). This diterpenediol showed substantial antispasmodic [2] and antimicrobial (unpublished work) activity. Further investigation of the petrol extract of *T. riparia* for minor constituents resulted in the isolation and characterization of an isomeric diterpenediol, i.e. 8(14),15-sandaracopimaradiene-2α,18-diol (2).

RESULTS AND DISCUSSION

The identity of 2 as a diterpenedial was immediately clear from its mass spectrum, which revealed a molecular ion at m/z 304 and a fragmentation pattern almost identical to that of diterpenedial 1 [1].

Comparison of the 360 MHz ¹H NMR spectra of diterpenediol 1 and the new natural diterpenediol indicated that the same substitution pattern was present except for the absence of hydroxylation at C-7. The characteristic three methyl singlets, the hydroxymethyl group at C-4, the position of the endocyclic olefinic double bond and the vinyl substituent led to the conclusion that the diterpene compound was substituted with a hydroxyl group on the ring skeleton of the diterpenediol. The typical splitting pattern of the hydroxyl-substituted methine moiety, characteristic for a proton having two methylene substituents, established the presence of a hydroxyl group at C-2.

The unambiguous assignments by ¹H NMR (360 MHz) of 27 of the 32 protons of 8(14),15-sandaracopimaradiene-2α,18-diol (2) was achieved using a combination of one-dimensional (1D) and two-dimensional (2D) proton NMR techniques, such as COSY [3] and delayed COSY [4]. The results are presented in Table 1.

The vicinal coupling constant (³J) of 11.5 and 4.1 Hz observed along the 1-2 and 2-3 fragments of the A-ring

are typical for an equatorial 2-(α)-hydroxy substituent in a rigid chair conformation [5]. The chair form of the B-ring with an exocyclic double bond is indicated by (a) the value of -14.2 Hz of the geminal coupling (2J) of the allylic 7-CH₂ group [6], (b) the vicinal axial-equatorial coupling 3J (ae) of 4.2 Hz and 3J (ee) of 2.0 Hz along the 6-7 bond, and (c) the cross-peaks of H-14 to H-9 and H-7 α in the COSY-2D spectra, but not to H-7 β which resides in the plane of the double bond [7]. The (averaged) vicinal couplings along the 9-11 bond are consistent with a non-chair C-ring containing the double bond.

8(14),15-Sandaracopimaradiene-2α,18-diol (2) is a new natural product. It has, however, already been obtained as a chemical degradation product of the naturally occurring diterpenetetrol hallol (4), which was isolated from the New Zealand podocarp Podocarpus halii Kirk [8]. The tetrol 4 was chemically modified by periodic acid oxidation and subsequent olefination of the intermediate aldehyde into the diterpenediol 2 using methylene-triphenylphosphorane. The data for the natural and

Table 1. ¹H NMR data for compound 2 (360 MHz, CDCl₃, TMS as internal standard)

Н	δ	Н	δ	H	δ
1α	1.006	7α	2.076	16	4.915
1 <i>β</i>	2.061	7β	2.26		4.890
2	3.917	9	1.837	17	1.046
3α	1.394	11	1.63	18	3.420
			1.53		3.159
3β	1.678	12	1.3-1.6	19	0.851
3β 5	1.2-1.5	14	5.258	20	0.883
6α	1.49	15	5.775		
6β	1.30				

J(Hz): $1\alpha,1\beta = -11.8$; $1\alpha,2 = 11.5$; $1\beta,2 = 4.1$; $1\beta,3\beta = 2.3$; $2,3\alpha = 11.5$; $2,3\beta = 4.1$; $3\alpha,3\beta = -12.0$; $6\alpha,7\beta = 2.0$; $6\beta,7\beta = 4.2$; $7\alpha,7\beta = -14.2$; $9,11\alpha = \sim 7.7$; $9,11\beta = \sim 7.7$; 15,16 = 17.5, 10.6; 16,16' = 1.4; 18,18' = 10.8.

^{*}To whom correspondence should be addressed.

Table 2. Comparison of the 13 C NMR data of the natural and synthetic diterpenediol (δ , CDCl₃)

C	Natural product	Synthetic product [8]	
<u> </u>	(this paper)		
1	48.2	48.1	
2	65.3	65.1	
3	44.7	44.7	
4	39.5*	39.4†	
5	47.0	47.0	
6	22.1	22.1	
7	35.5	35.5	
8	136.2	136.2	
9	50.5	50.4	
10	39.7*	39.6†	
11	18.86‡	18.9§	
12	34.4	34.4	
13	37.0	36.9	
14	129.5	129.3	
15	148.9	148.7	
16	110.2	110.0	
17	25.9	25.9	
18	71.7	71.5	
19	18.95‡	18.9§	
20	16.5	16.5	

^{*,†,‡,§}Assignments may be interchanged.

synthetic 2 are in good agreement, although our ¹H NMR data (360 MHz) and mass spectral data are more complete. Final proof of the common identity of the natural compound and the synthetic compound came from a comparison of the ¹³C NMR data (20 MHz), which were almost identical (Table 2).

EXPERIMENTAL

4 (Hallol)

Plant material. Leaves of T. riparia were collected in the prefecture of Butare (South-West of Rwanda) in October 1984 and identified as previously described.

Extraction and isolation. Air-dried leaves (522 g) were ground to a fine powder, which was extracted in a percolator with petrol (40-60°) (12 l.). The extract was filtered, concentrated in vacuo and extracted with MeOH-H₂O (9:1). The petrol phase was then evaporated in vacuo to give a brown-green syrup (14.9 g, 2.8%), which was extracted with MeOH-H₂O. After evaporation of MeOH, the aq. phase was extracted with CHCl₃, which gave, after evaporation in vacuo, a brown residue (24.8 g, yield 4.5%). 12 g of the CHCl₃ extract was adsorbed on silica gel (50 g) and slurried onto the top of a column containing 550 g silica gel (Riedel de Haën, 230-400 mesh) in n-hexane and eluted with a n-hexane-toluene-CHCl₃-EtOAc-MeOH gradient.

8(14),15-Sandaracopimaradiene- 2α ,18-diol (1) was isolated from the EtOAc fraction (141 mg, yield 0.05%) and was recrystallized from cyclohexane to afford small white crystals, mp $203-204^{\circ}$, $[\alpha]_D^{20}-21.7^{\circ}$ (c 0.35; CHCl₃) (589 nm). ¹H NMR (360 MHz) and ¹³C NMR (20 MHz): see Tables 1 and 2; IR (KBr): 3600-3100 cm⁻¹ (ν_{OH} , broad); MS m/z (rel. int.): 304 [M]⁺ (9), 286 (40), 256 (33), 187 (100), 241 (16), 187 (100), 159 (41), 132 (32), 121 (70), 119 (51), 111 (27), 109 (32), 107 (41), 105 (43), 96 (36), 94 (45), 92 (45), 90 (41), 83 (36), 81 (58), 79 (34), 73 (30), 71 (30), 69 (51), 67 (32), 57 (55), 55 (72).

Acknowledgement—The authors are indebted to the Belgian Nationaal Fonds voor Wetenschappelijk Onderzoek (National Fund for Scientific Research) for financial support.

REFERENCES

- De Kimpe, N., Schamp, N., Van Puyvelde, L., Dubé, S., Chagnon-Dubé, M., Borremans, F., Anteunis, M. J. O., Declercq, J.-P., Germain, G. and Van Meerssche, M. (1982) J. Org. Chem. 47, 3628.
- 2. Van Puyvelde, L., De Kimpe, N., Schamp, N., Lefebvre, R. and

- Mugabo, P. (1984) Farm. Tijdschr. Belg. 61, 317.
- 3. Aue, W. P., Bartholdi, E. and Ernst, R. R. (1976) J. Chem. Phys. 64, 2229.
- 4. Bax, A. and Freeman, R. (1981) J. Magn. Reson. 44, 542.
- Haasnoot, C. A. G., de Leeuw, F. A. A. M. and Altona, C. (1980) Tetrahedron 36, 2783.
- Barfield, M., Hruby, V. J. and Meraldi, J. P. (1976) J. Am. Chem. Soc. 98, 1038.
- Barfield, M., Spear, R. J. and Sternhell, S. (1976) Chem. Rev. 76, 593.
- 8. Cambie, R. C., Burfitt, I. R., Goodwin, T. E. and Wenkert, E. (1975) J. Org. Chem. 40, 3789.